

# Emulsification

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**“Physics of droplets: Basic and  
advanced topics”**

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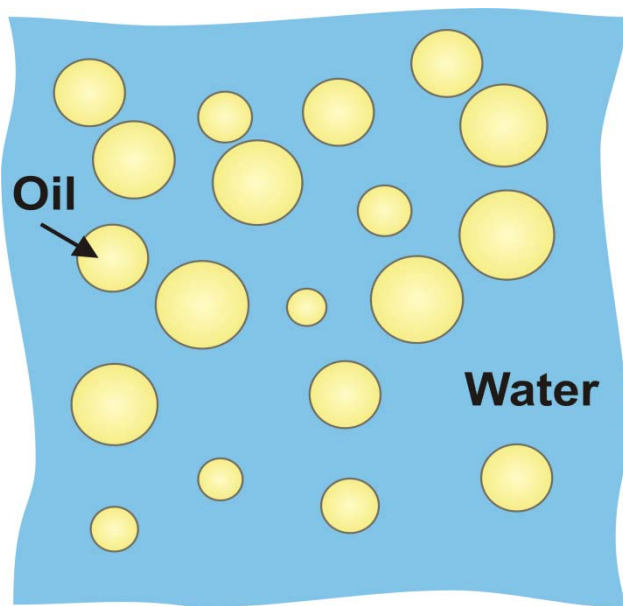
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- 2. Processes during emulsification.**
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# Types and applications of emulsion

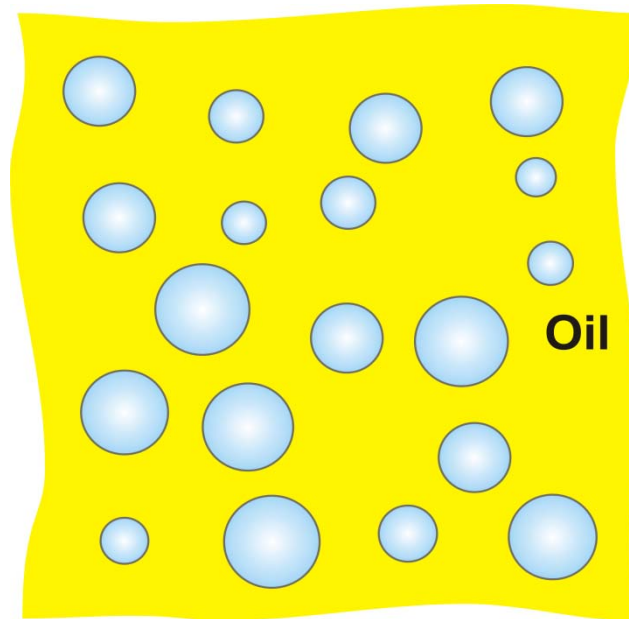
## Emulsion types

Oil-in-Water (O/W)



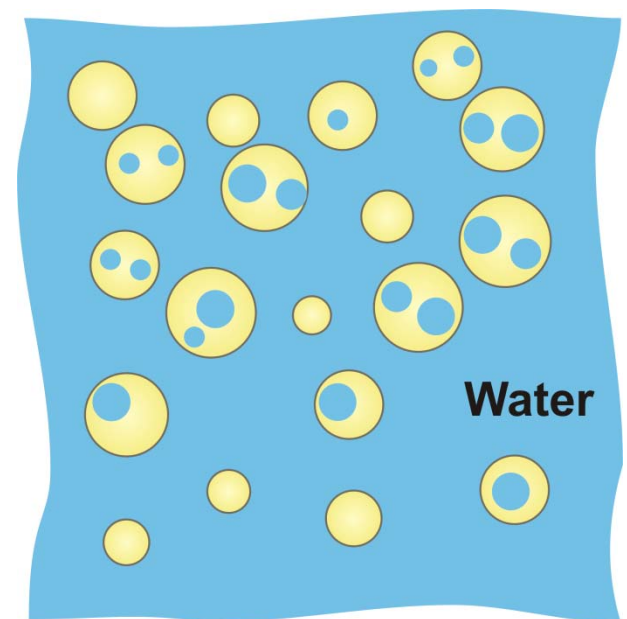
Direct emulsion

Water-in-oil (W/O)



Inverse emulsion

W/O/W



Double emulsion

# Cosmetic and pharmaceutical products



**Creams, Lotions, Conditioners, Body wash, ...**

# Food and beverages



**Milk**  
**Mayonnaise**  
**Margarine**  
**Butter**  
**Cheese**

...

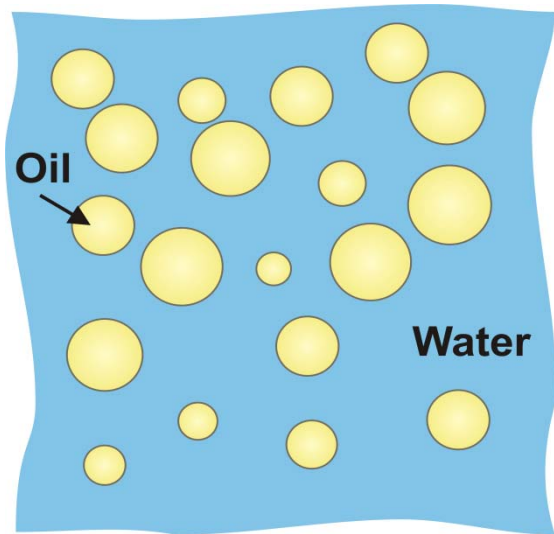
# Oil recovery and processing



**Both W/O and O/W emulsions are often encountered**

# Main characteristics of the emulsions

## Oil-in-Water (O/W)



**Oil volume fraction**

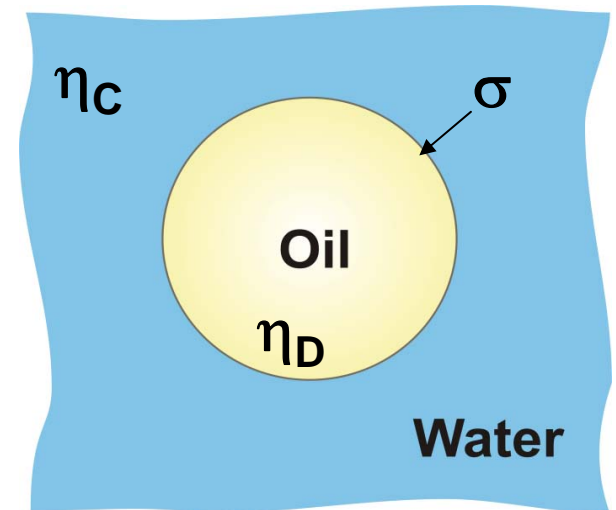
$$\Phi = \frac{V_{OIL}}{V_{EM}}$$

**Mean volume-surface radius**

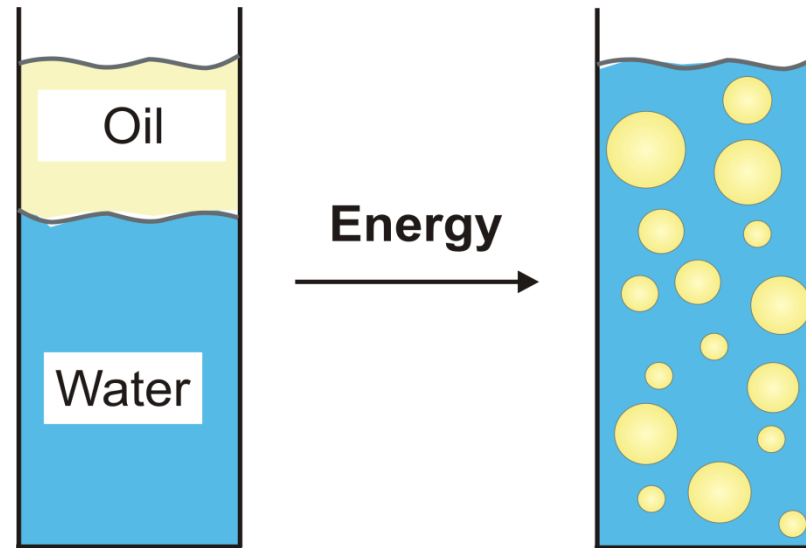
$$R_{32} = \frac{\sum N_i R_i^3}{\sum N_i R_i^2}$$

**Direct emulsion**

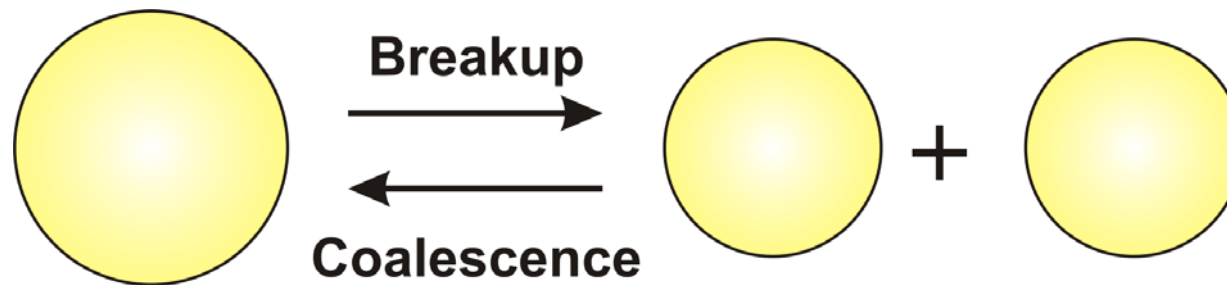
- Viscosity of dispersed phase,  $\eta_D$
- Viscosity of continuous phase,  $\eta_C$
- Interfacial tension,  $\sigma$



# Emulsification

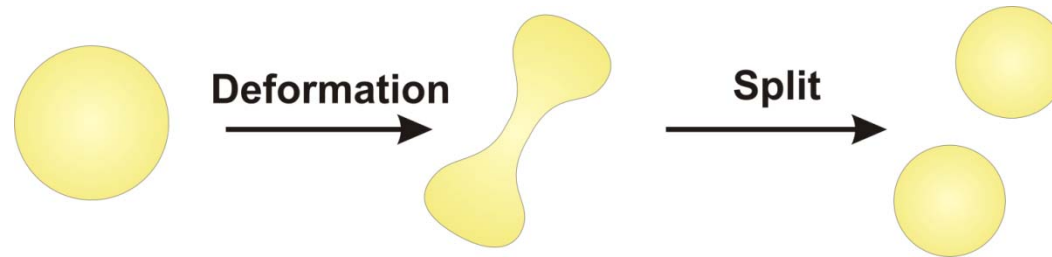


## Processes during emulsification





# Requirements for drop breakup



## Pressure balance, Davies 1985

Applied stress = Capillary pressure + Viscous stress inside drop

$$\tau \sim \sigma/d + \eta_D \left[ \delta \langle u_d \rangle / \delta x \right]$$

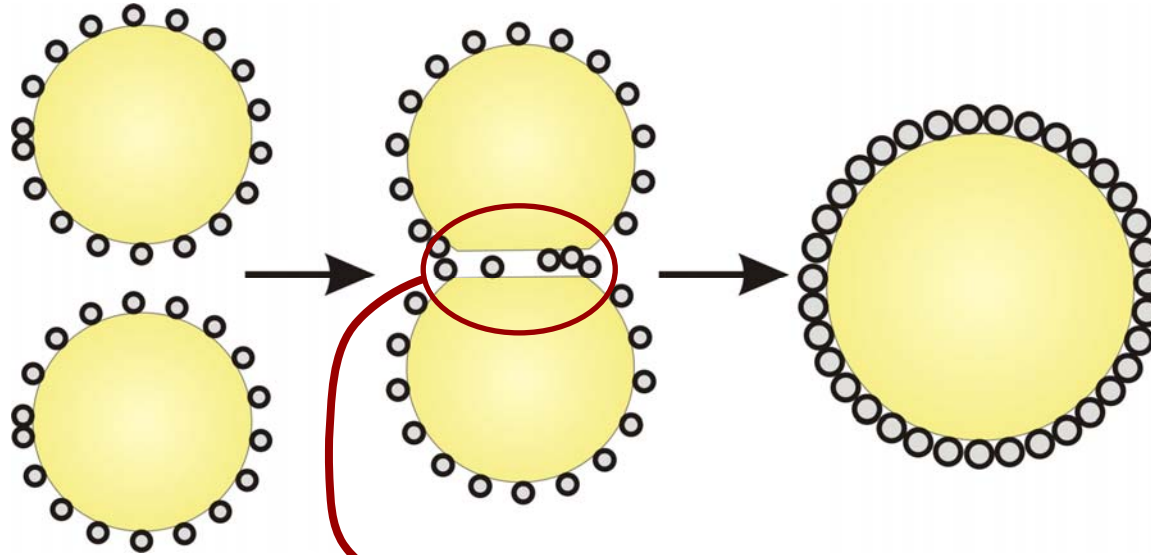
## Time scales (Walstra, 1983)

Deformation time,  $t_{\text{DEF}}$

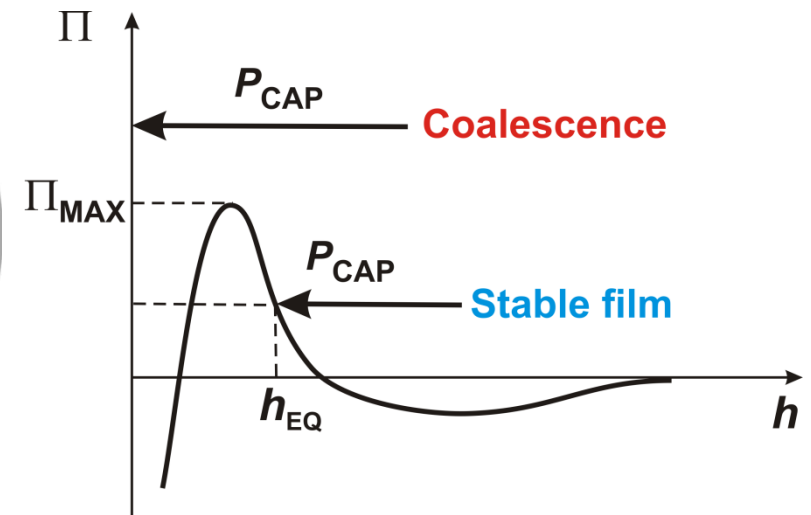
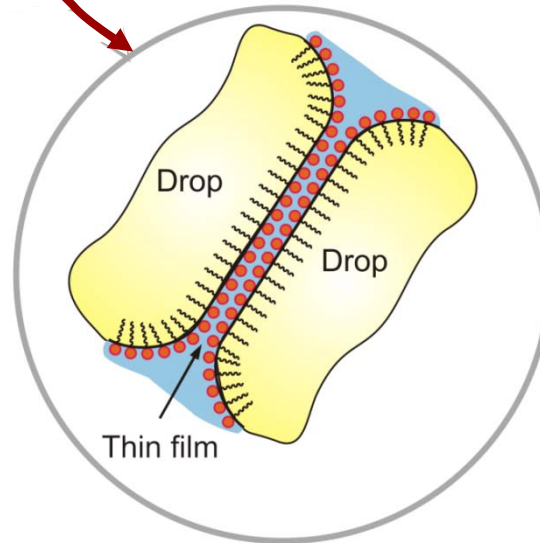
Characteristic time for applied stress

$$t_{\text{DEF}} < t_c$$

# Drop-drop coalescence

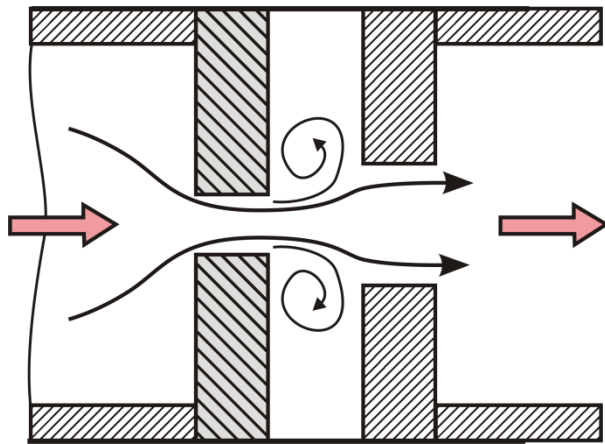


**Stability of the emulsion film is determined by the surfactants adsorbed**

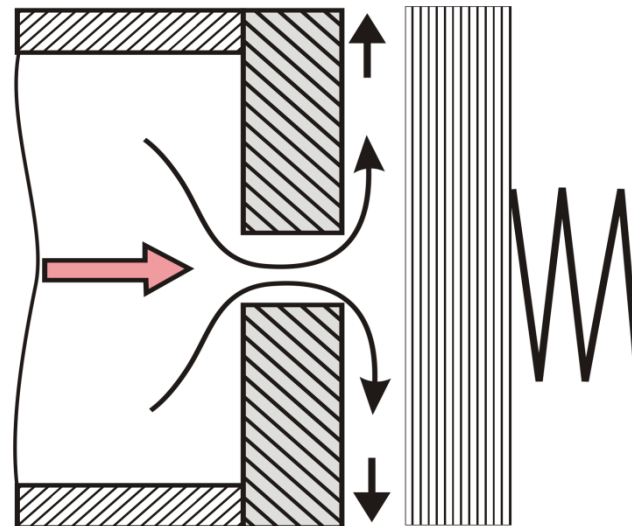
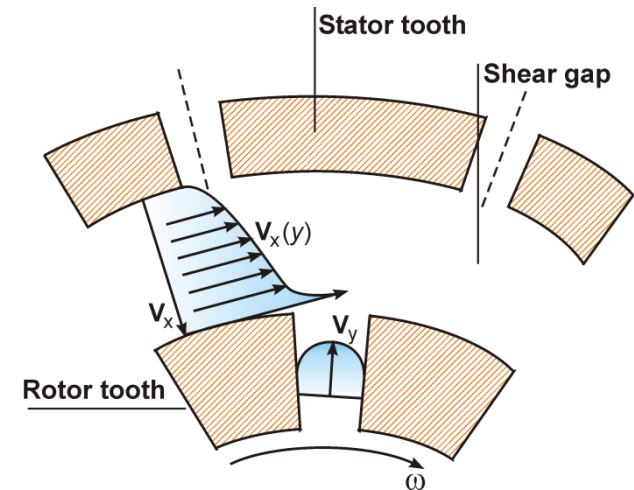


# Types of apparatuses

## Microfluidizer

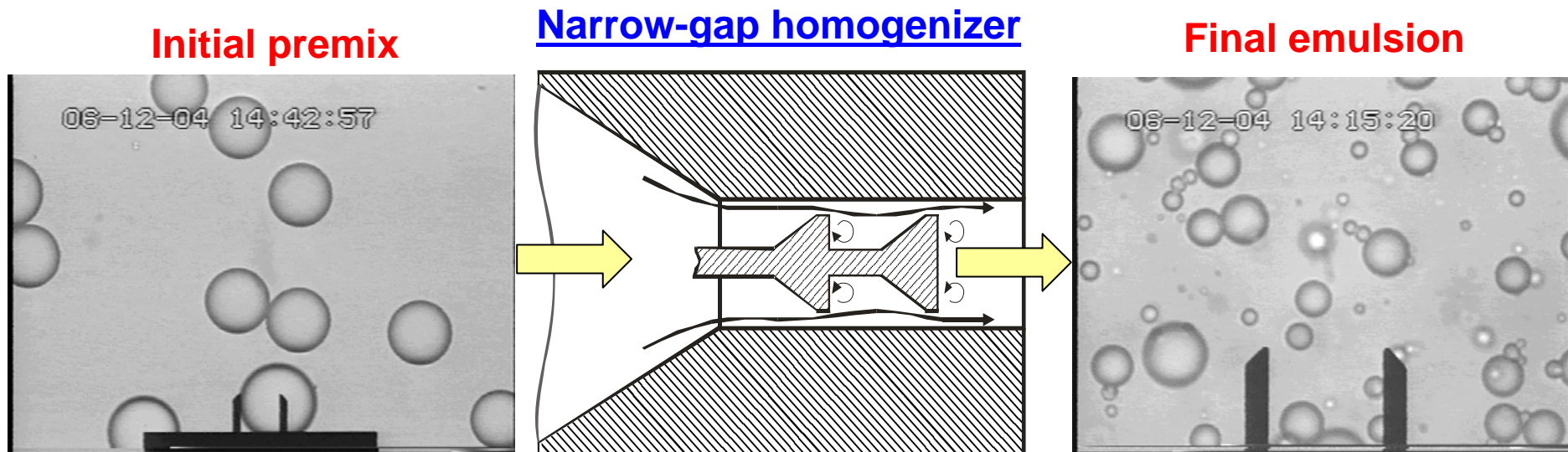


## Rotor-stator homogenizer



## Valve homogenizer

# Emulsification method: narrow gap homogenizer

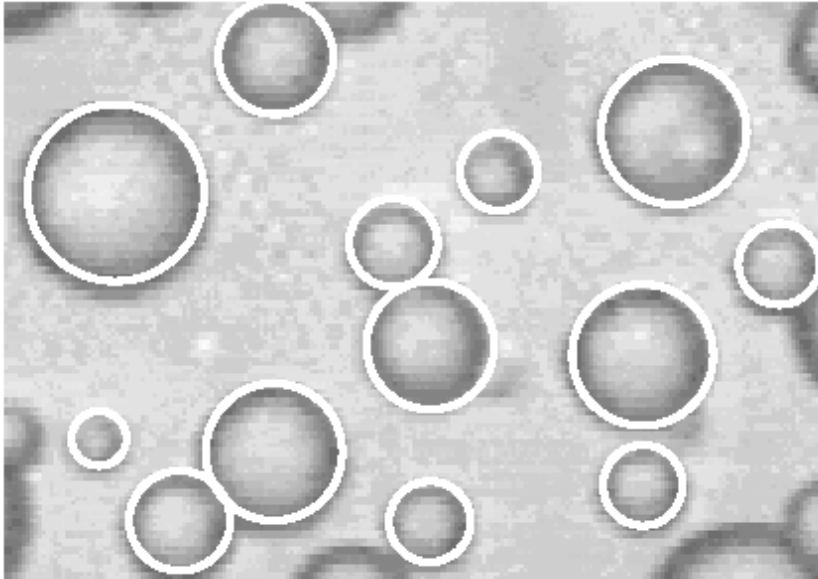


**Main advantage - well defined hydrodynamic conditions**

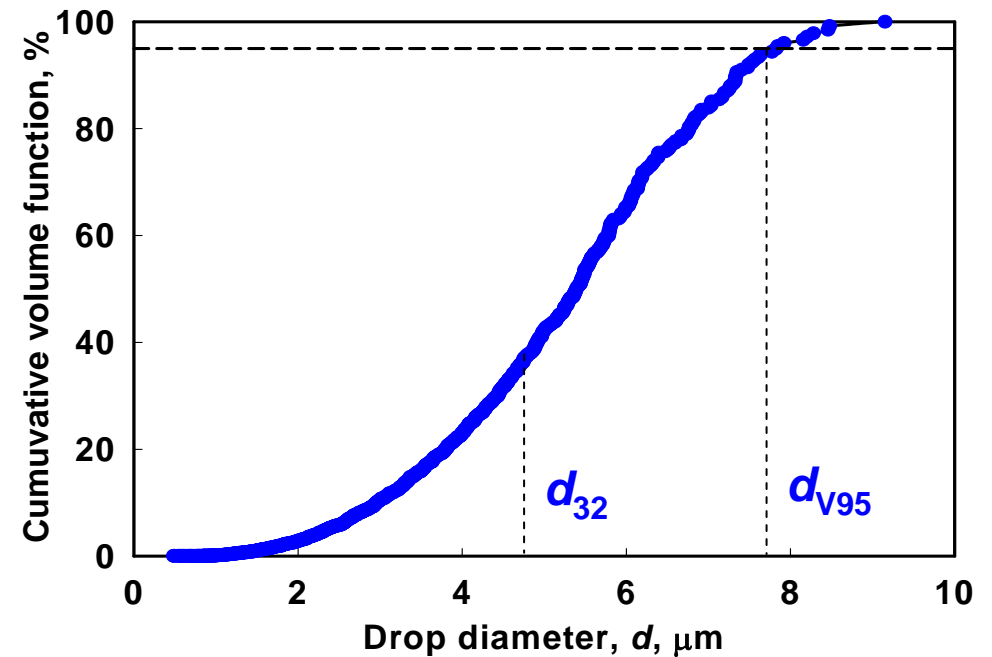
- Rate of energy dissipation per unit mass,  $\varepsilon$
- Residence time,  $\theta$

# Drop-size distribution

## Optical microscopy



## Maximum diameter, $d_{V95}$

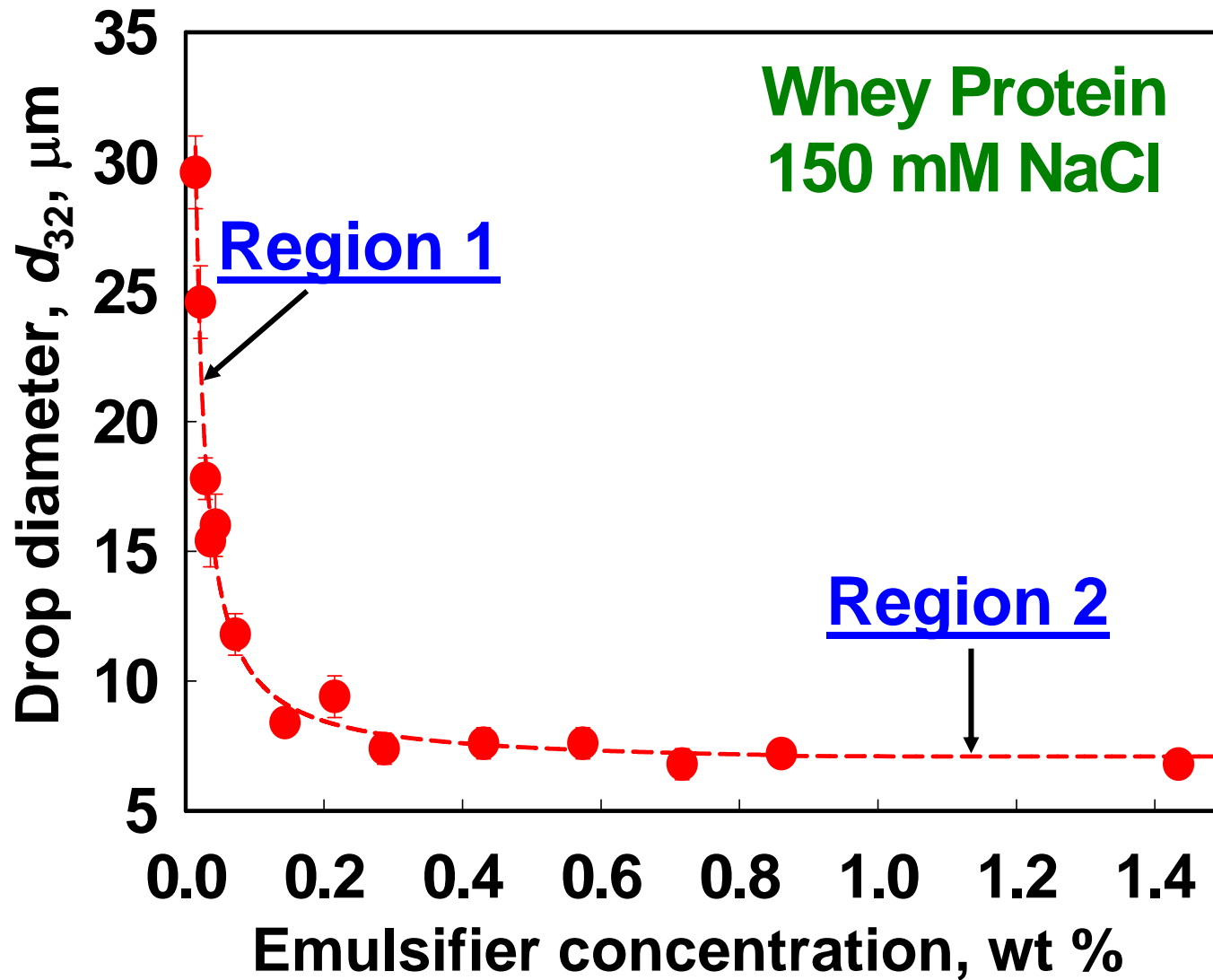


## Mean volume-surface diameter, $d_{32}$

$$d_{32} = \frac{\sum N_i d_i^3}{\sum N_i d_i^2}$$

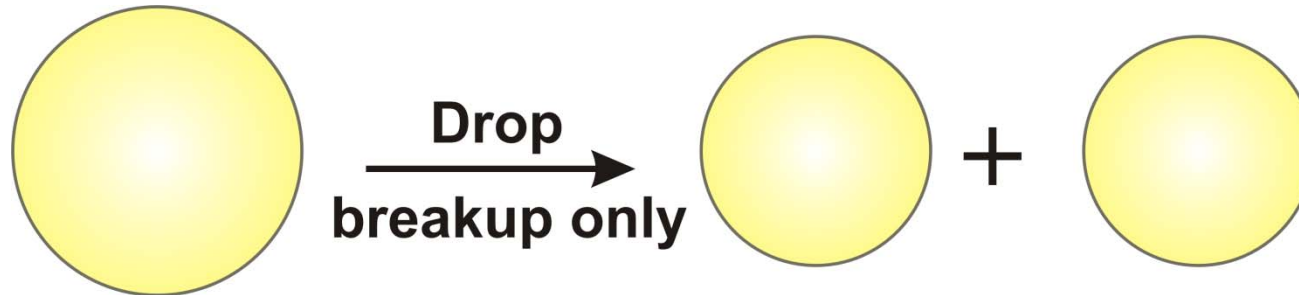
**Main advantage – very precise characterization**

# Effect of emulsifier concentration



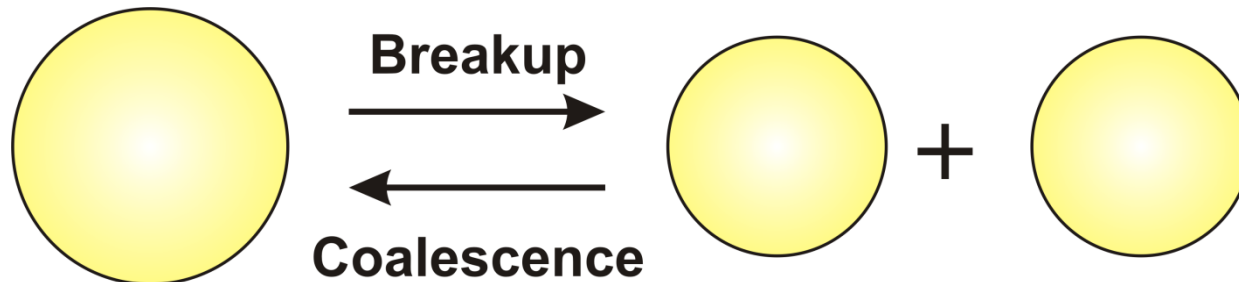
## High emulsifier concentration

- Negligible coalescence
- $d_{32}$  is determined by drop breakup only



## Low emulsifier concentration

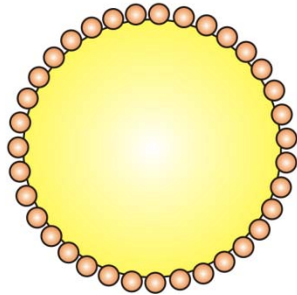
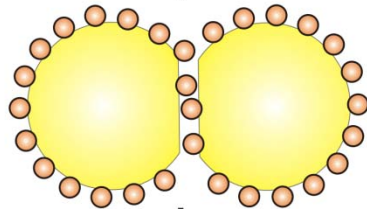
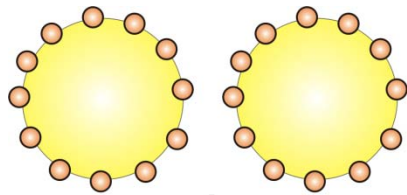
- Significant effect of drop coalescence on  $d_{32}$



# Emulsification with coalescence

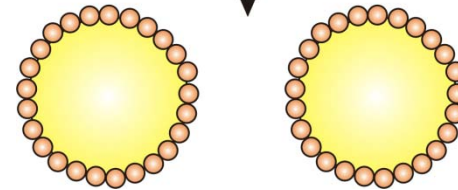
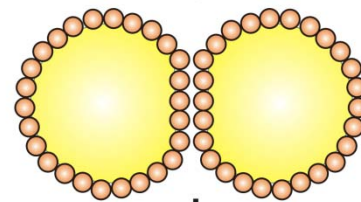
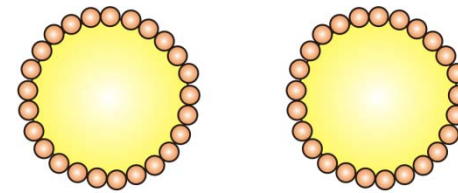
## (A) No electrostatic repulsion

Monolayer adsorption  $\Gamma_M$  needed to stabilize the drops



Coalescence

$$\Gamma < \Gamma_M$$



No coalescence

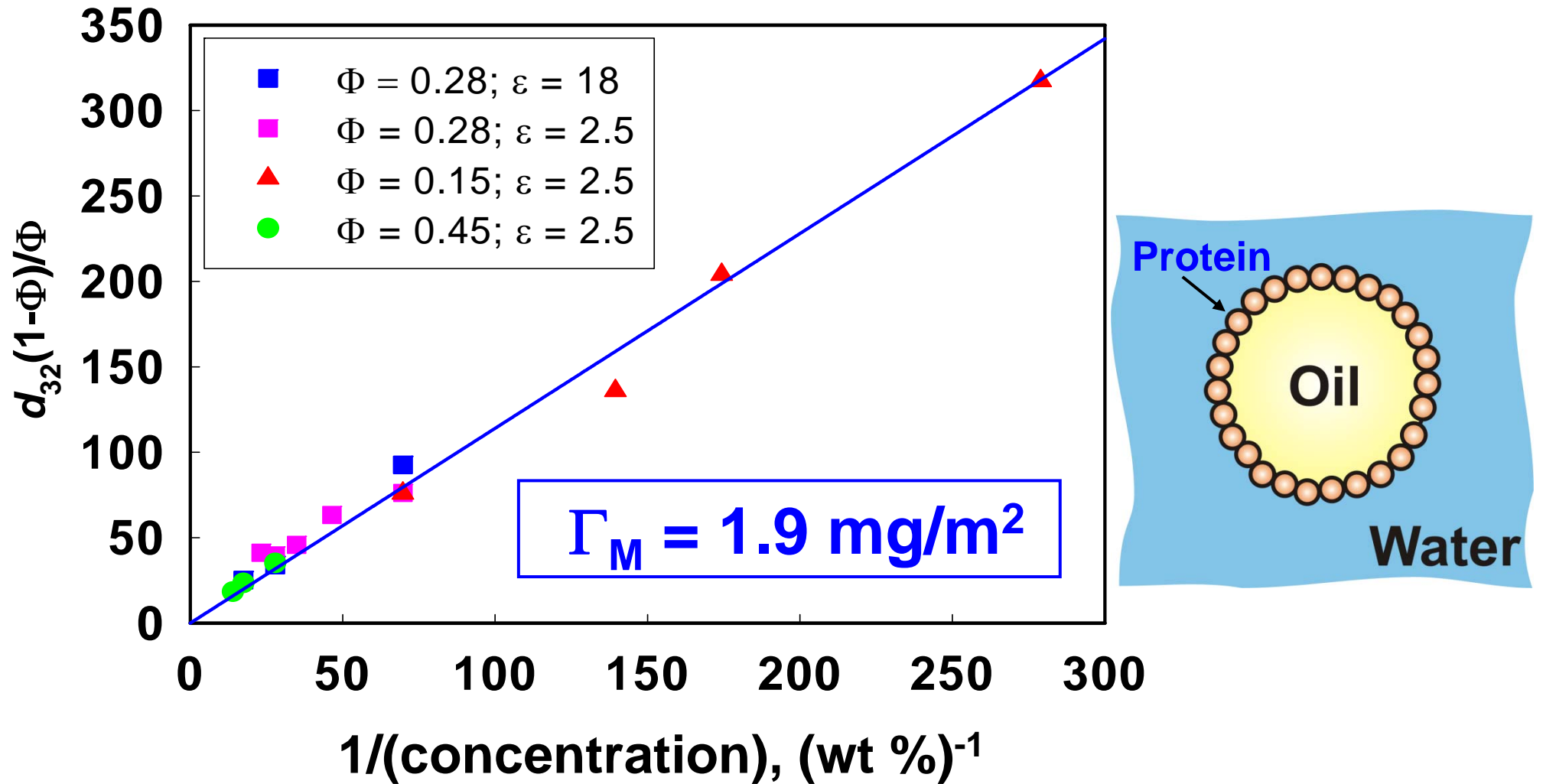
$$\Gamma \geq \Gamma_M$$

$\Rightarrow$  Drop size:

$$d_{32} \approx \frac{6\Phi}{10(1-\Phi)} \frac{\Gamma_M}{C_{INI}}$$

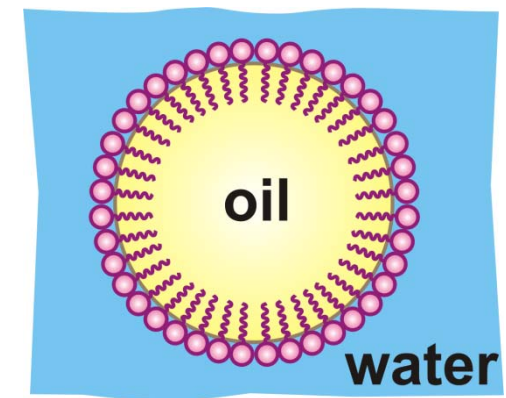
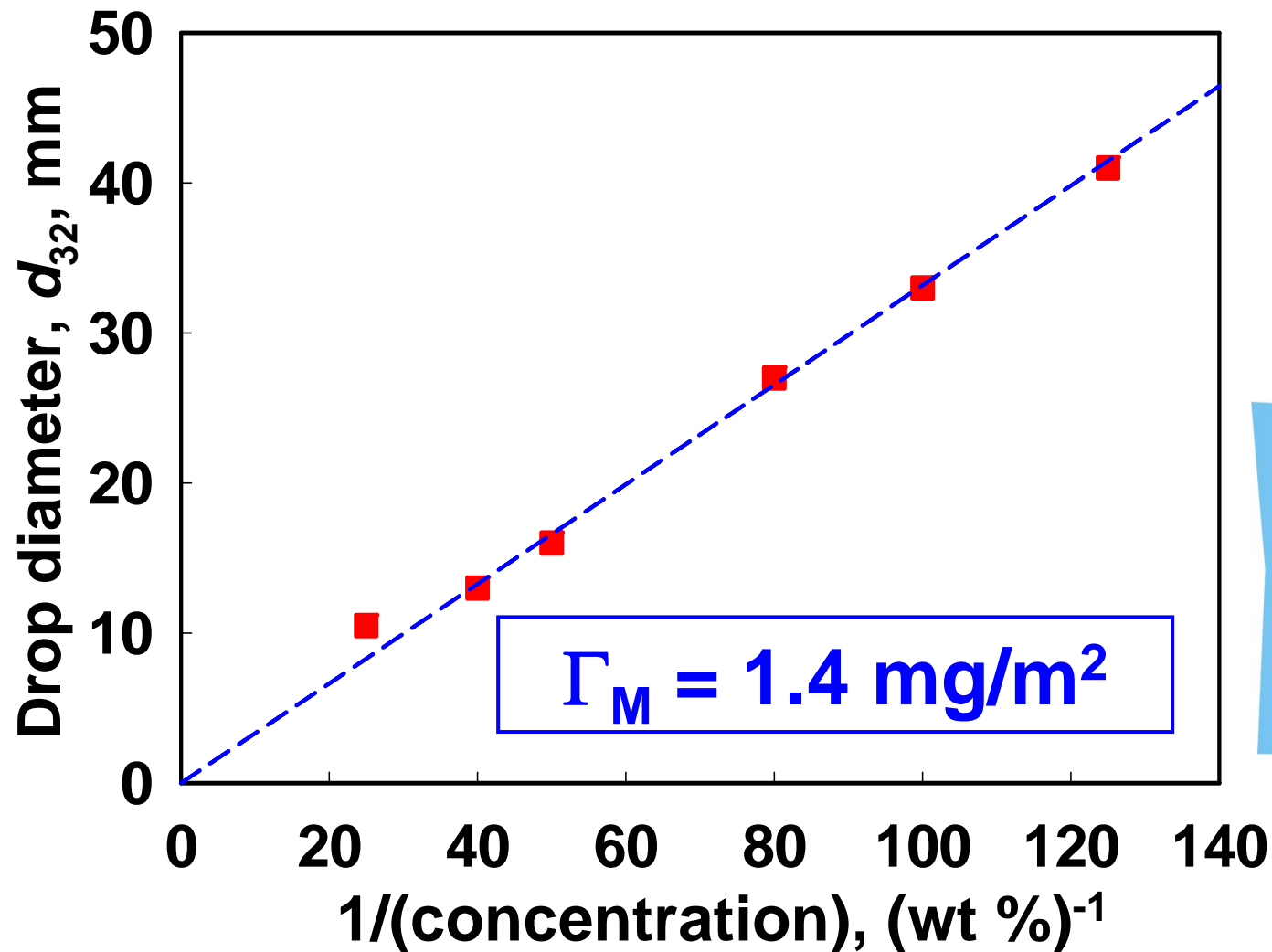


# Comparison with experimental data (whey protein + 150 mM NaCl)



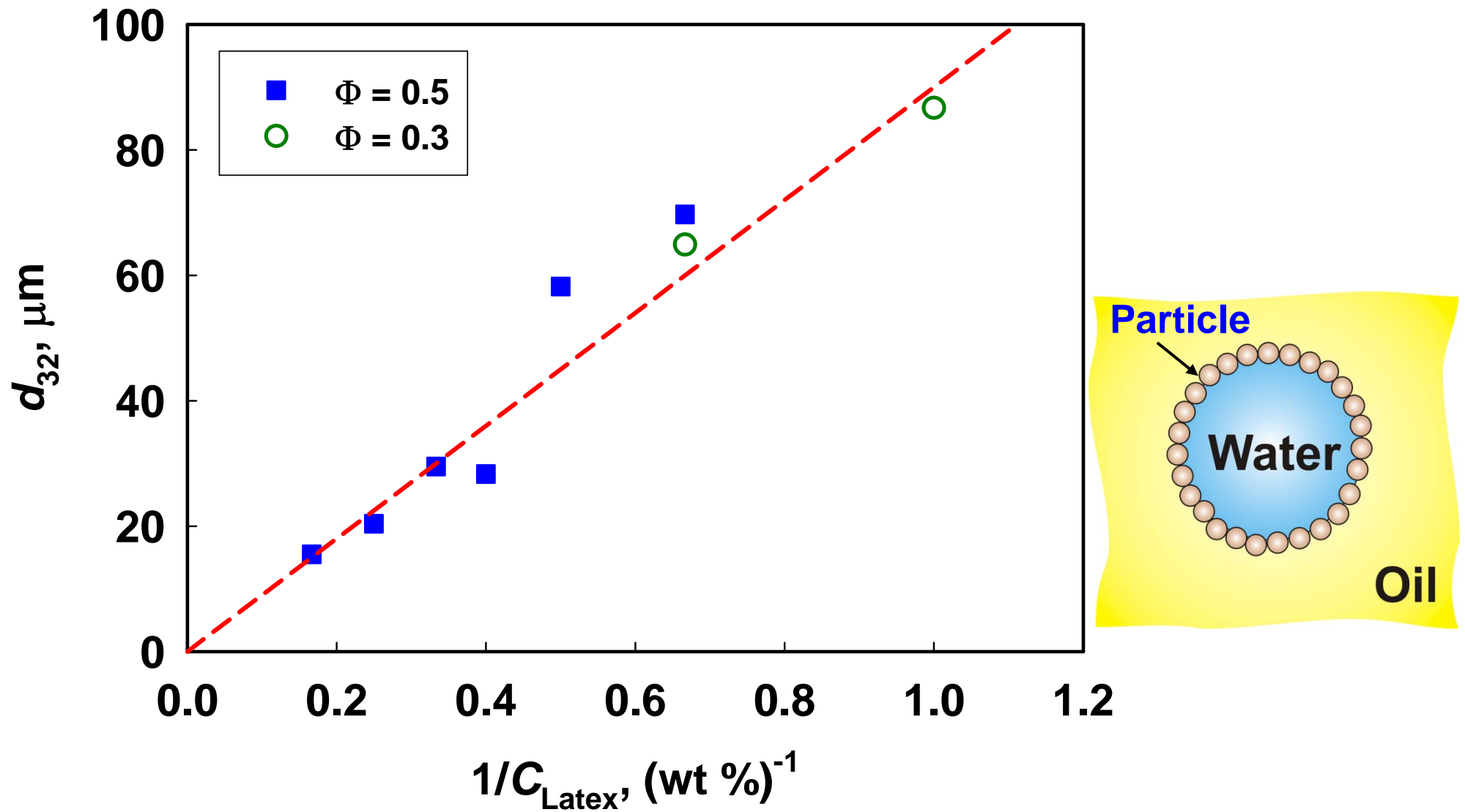
Tcholakova et al, *Langmuir*, 2003, 19, 5640; *Langmuir*, 2004, 20, 7444;

# Nonionic surfactant Brij 58 + 150 mM NaCl



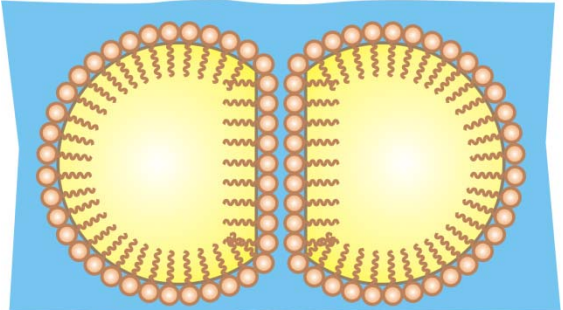
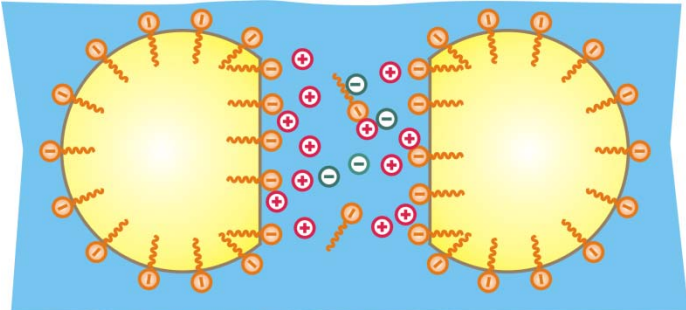
Tcholakova et al, *Langmuir*, 2004, 20, 7444; Tcholakova et al, *PCCP*, 2008, 10, 1608.

# Solid latex particles + 500 mM NaCl



Golemanov et al, *Langmuir*, 2006, 22, 4698; Tcholakova et al, *PCCP*, 2008, 10, 1608.

# Degree of coverage, $\theta = \Gamma / \Gamma_M$ , preventing coalescence

System	Theory	Experimental	
WPC 150 mM NaCl	$\geq 1$	$\geq 1$	<p><b>Steric repulsion</b></p> 
Brij 58 150 mM NaCl	$\approx 1$	$\approx 1$	
SDS 10 mM NaCl	$\approx 0$	$< 0.05$	<p><b>Electrostatic repulsion</b></p> 
SDS 150 mM NaCl	$\approx 0.23$	$\approx 0.3$	

S. Tcholakova; N. Denkov and T. Danner, *Langmuir*, 2004, 20, 7444.

# Conclusions

– emulsification with coalescence

## A. Suppressed electrostatic repulsion

The model with  $\Gamma_M$  describes very well the data !

⇒ Typical for nonionic surfactants, solid particles and proteins at high electrolyte concentration.

## B. Significant electrostatic repulsion

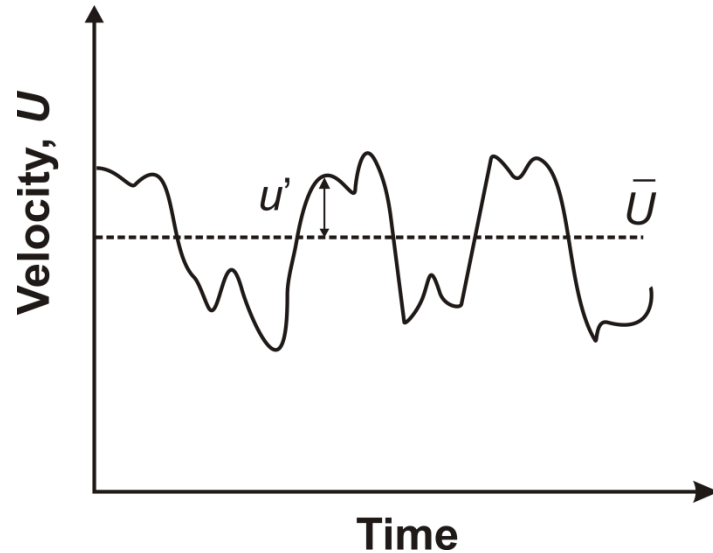
DLVO theory describes well the data !

⇒ Typical for ionic surfactants and proteins.

# **Emulsification in turbulent flow without coalescence**

- 1. Main characteristics of turbulent flow**
  - Velocity fluctuation**
  - Rate of energy dissipation**
- 2. Emulsification in inertial regime**
- 3. Emulsification in viscous regime**

# Main characteristics of turbulent flow



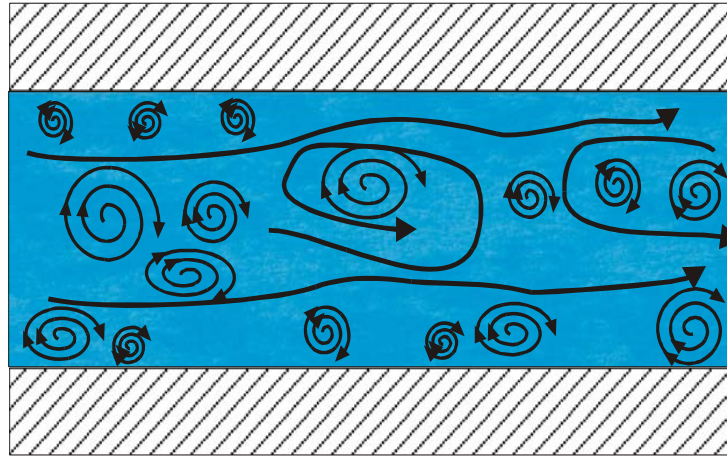
## Velocity fluctuation

$$u = \left\langle (U - \bar{U})^2 \right\rangle^{1/2}$$

## Rate of energy dissipation

$$\varepsilon \sim \frac{\dot{E}_{KIN}}{m} \sim \frac{m u^2}{m(l/u)} \sim \frac{u^3}{l}$$

# Size of the eddies in the turbulent flow



Largest eddies  $\approx$  diameter of the pipe

## The smallest eddies

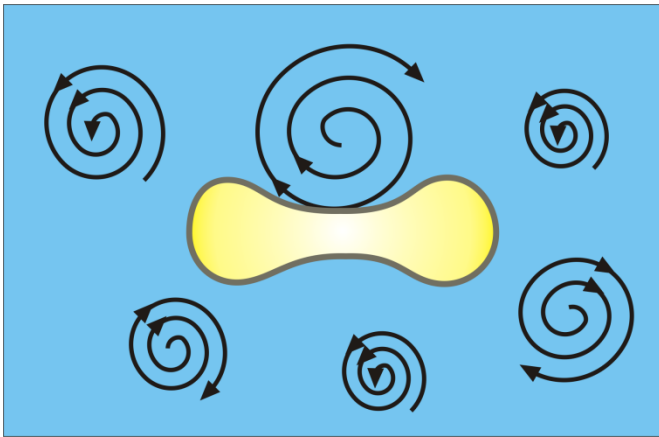
$$\text{Re} \sim \frac{H u \rho_c}{\eta_c} \sim 1$$

$$\lambda_0 = \varepsilon^{-1/4} \eta_c^{3/4} \rho^{-3/4}$$



# Drops in the turbulent flow

## Inertial regime

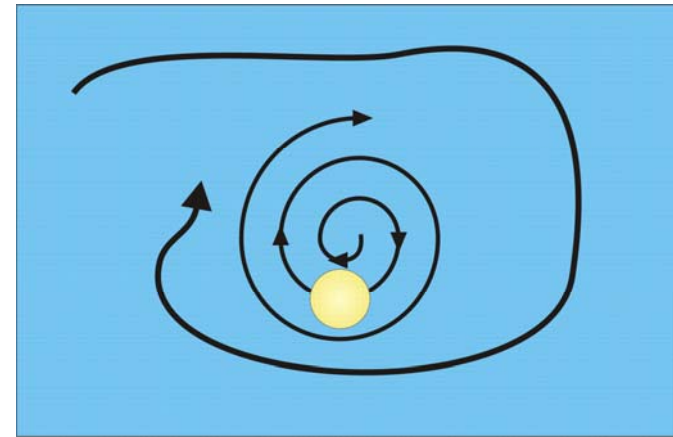


$$d > \lambda_0$$

Pressure  
fluctuations

$$\rho \langle u \rangle^2$$

## Viscous regime



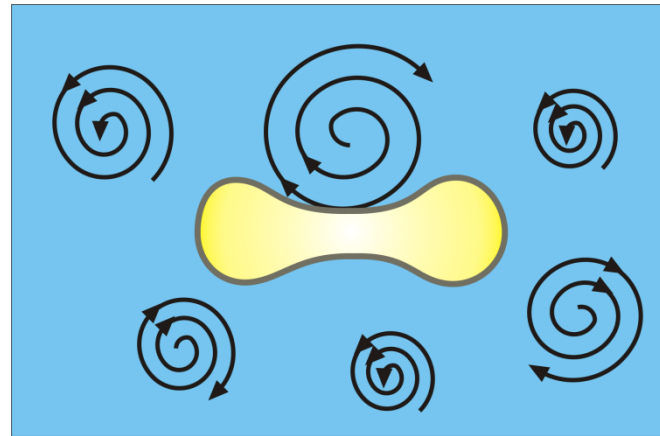
$$d < \lambda_0$$

Viscous  
stresses inside  
smallest eddies

$$\eta_c (dU/dx)$$

# Drop breakup in inertial turbulent regime

(Kolmogorov, Hinze; Davies, Calabrese)



Pressure  
fluctuations

=

Capillary  
pressure

+

Viscous stress  
inside drop

$$\rho \langle u \rangle^2$$

~

$$\sigma/d$$

+

$$\eta_D \left[ \delta \langle u_d \rangle / \delta x \right]$$

$$\rho (\varepsilon d)^{2/3}$$

=

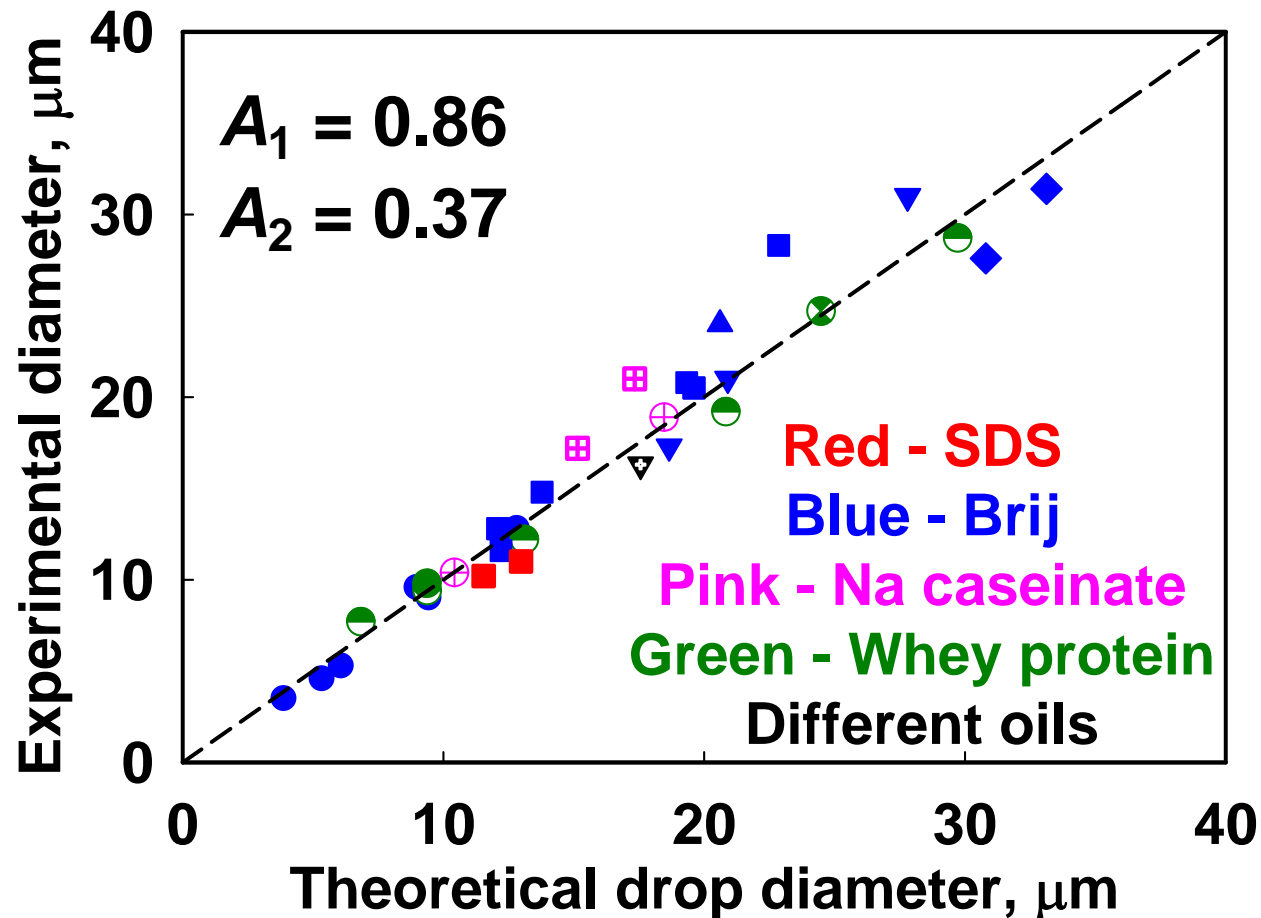
$$C_1 \sigma/d$$

+

$$C_2 \eta_D \varepsilon^{1/3} / d^{2/3}$$

# Predicted vs. measured drop diameter (inertial regime)

$$d = A_1 \left( \sigma + A_2 \eta_D \varepsilon^{1/3} d^{1/3} \right)^{3/5} \rho_C^{-3/5} \varepsilon^{-2/5}$$



# Effect of oil viscosity on emulsification in inertial regime

Oil Viscosity, $\eta_D$ , Pa.s	0.1	0.6	1.5	10	60	100
<b>1 wt % PVA</b> $\sigma = 21$ mN/m	<b><math>19 \pm 4</math> <math>\mu\text{m}</math></b>	<b><math>24 \pm 6</math> <math>\mu\text{m}</math></b>	<b>Millimeter sized drops + Non emulsified oil</b>			
<b>10 wt % SDS</b> $\sigma = 7.0$ mN/m	<b><math>10 \pm 2</math> <math>\mu\text{m}</math></b>	<b><math>13 \pm 3</math> <math>\mu\text{m}</math></b>				

## Time scales (Walstra, 1983)

**Deformation time**

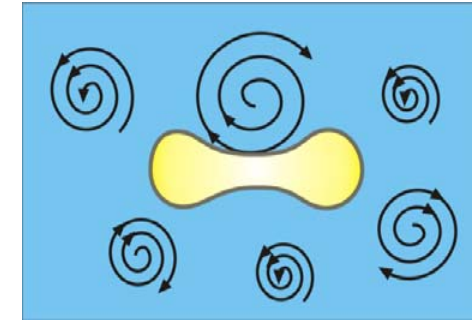
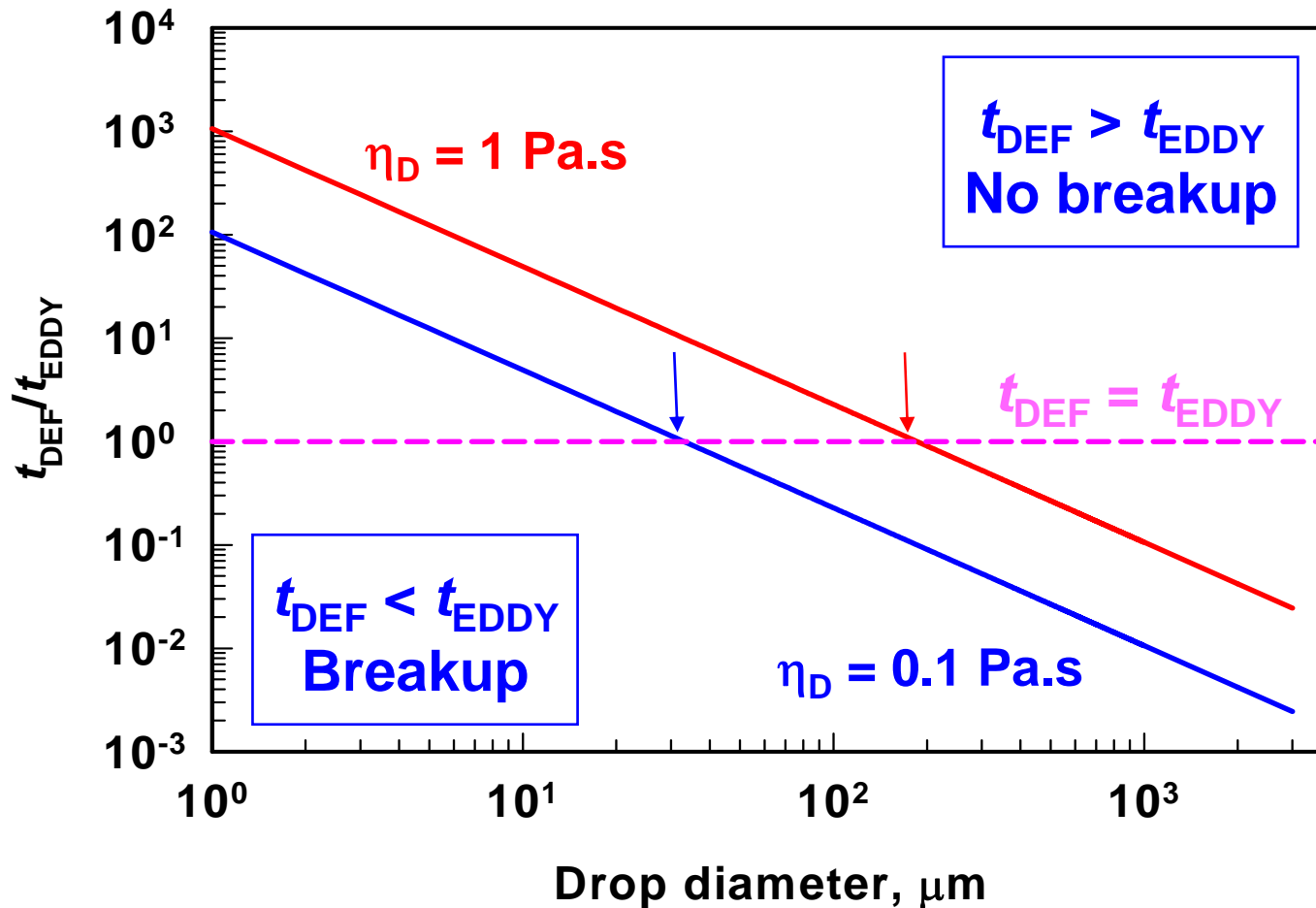
$$t_{DEF} = \frac{\eta_D}{5\varepsilon^{2/3} d^{2/3} \rho_C^{1/3}}$$

**Eddies life time**

$$t_{EDDY} = \frac{d^{2/3} \rho_C^{1/3}}{\varepsilon^{1/3}}$$

$$t_{DEF} < t_{EDDY} \quad \Rightarrow \quad t_{DEF}/t_{EDDY} < 1$$

# Deformation vs life time of eddies

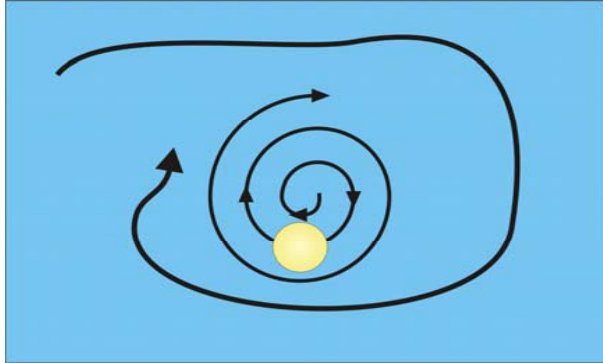


## Requirement for drop breakup

$\eta_D = 1 \text{ Pa.s } d > 270 \mu\text{m}$   
 $\eta_D = 0.1 \text{ Pa.s } d > 27 \mu\text{m}$

**Inertial regime of emulsification is unsuitable for oils with viscosity  $\geq 1 \text{ Pa.s}$ , because of too long deformation time!**

# Emulsification in viscous turbulent regime



## Maximal stable drop diameter

Kolmogorov, 1949; Hinze, 1955

$$d_V = A_3 \sigma / (\varepsilon \eta_C \rho_C)^{1/2}$$

## Deformation vs residence time

Walstra, 1983

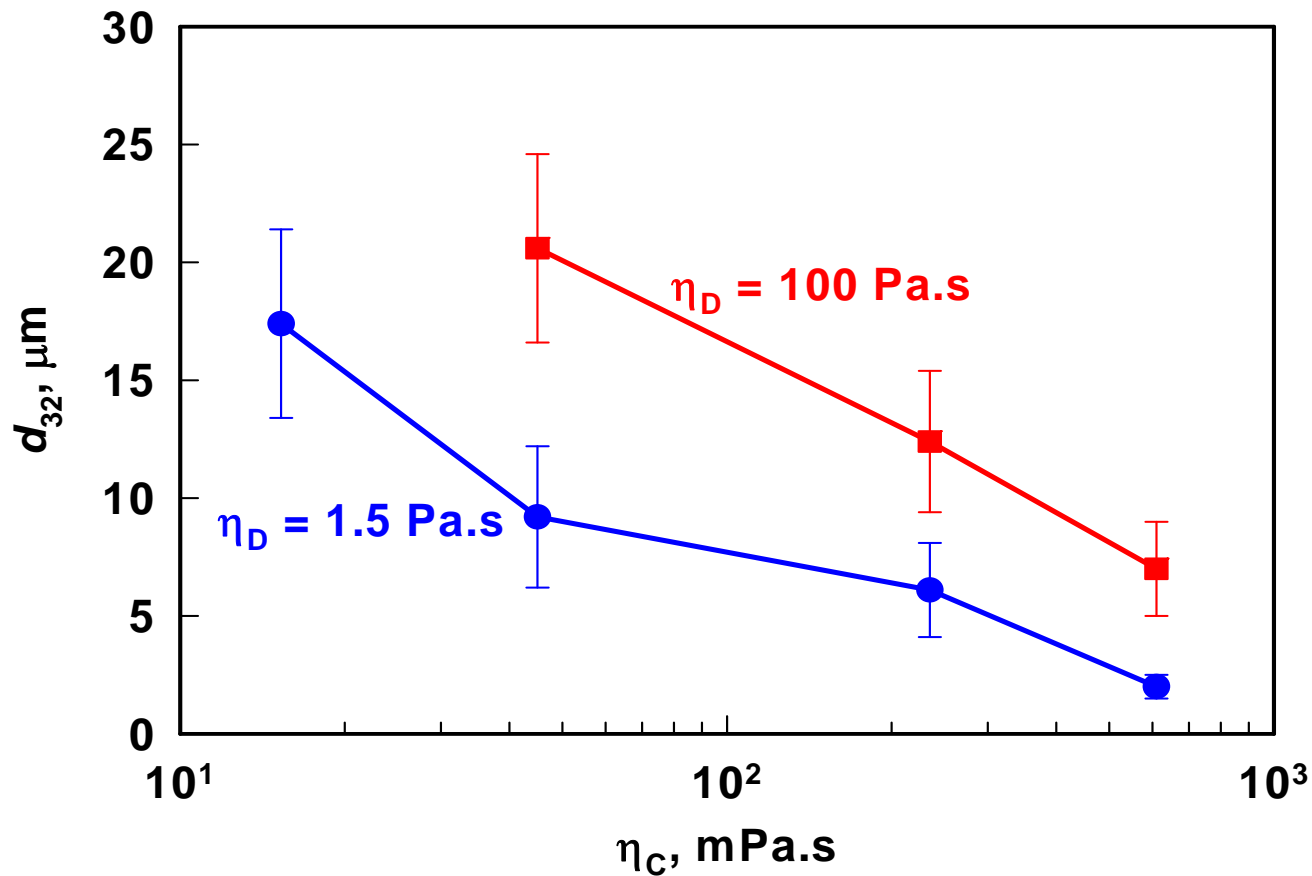
$$\tau_{DEF} = \frac{\eta_D}{(9 \varepsilon \eta_C)^{1/2}}$$

$$t_{RES} \approx 4 \text{ ms}$$

$$\eta_D = 10 \text{ Pa.s}; \varepsilon = 2 \times 10^8 \text{ J/m}^3 \cdot \text{s}; \rho_C = 10^3 \text{ kg/m}^3$$

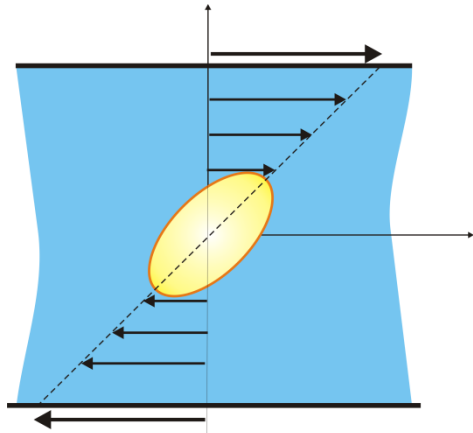
$$\Rightarrow t_{RES} > t_{DEF}, \text{ when } \eta_C > 35 \text{ mPa.s}$$

## Effect of solution and oil viscosity on mean drop size in viscous regime

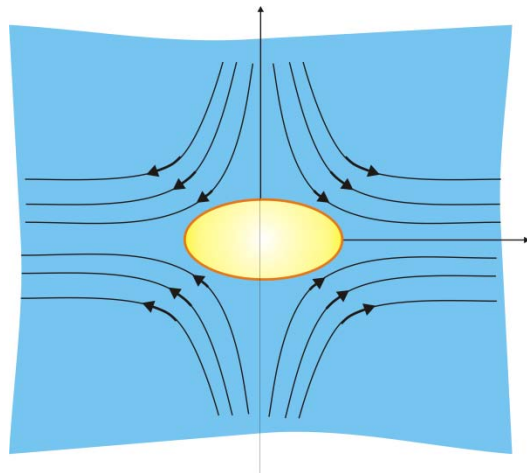


Successful emulsification even for oils with  $\eta_D \sim 100 \text{ Pa}\cdot\text{s}$

# Emulsification in Laminar flows



Simple shear,  $\alpha = 0$



Elongational,  $\alpha = 1$

S. Guido and co-authors, 2006



(1) Drop stretching

(2) Capillary instability

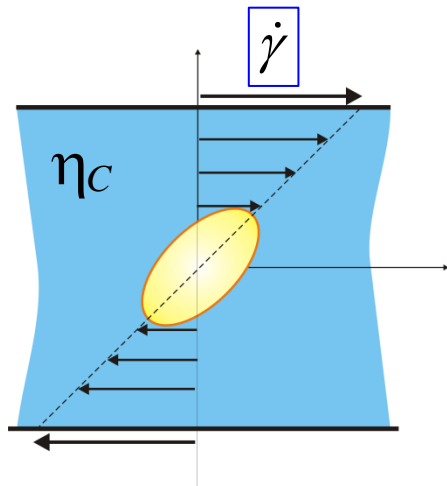
(3) Daughter drops of different sizes



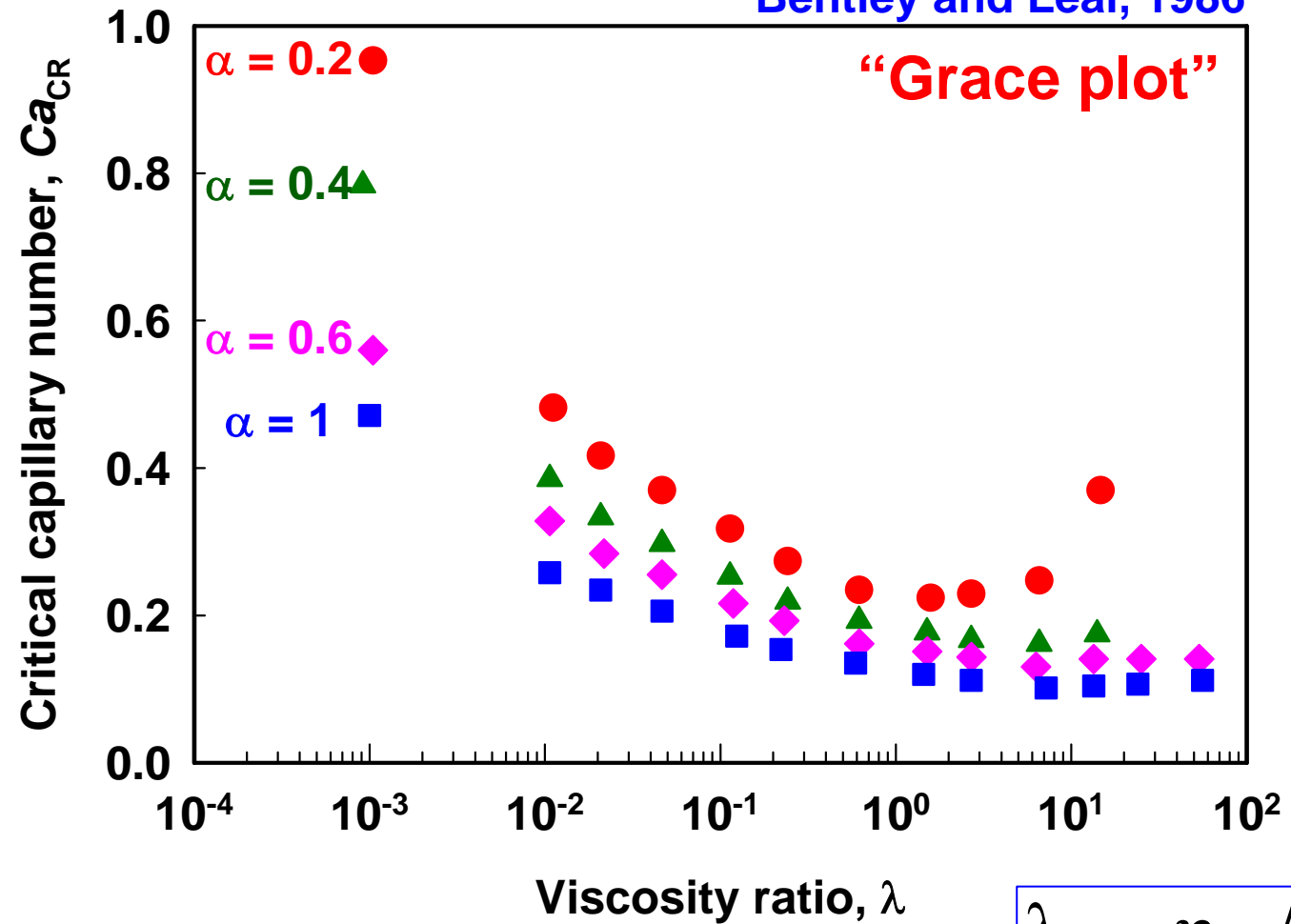
# Main factors controlling drop breakup in laminar flows

## Capillary number

$$Ca = \left( \frac{\eta_c \dot{\gamma}}{\sigma/R} \right)$$



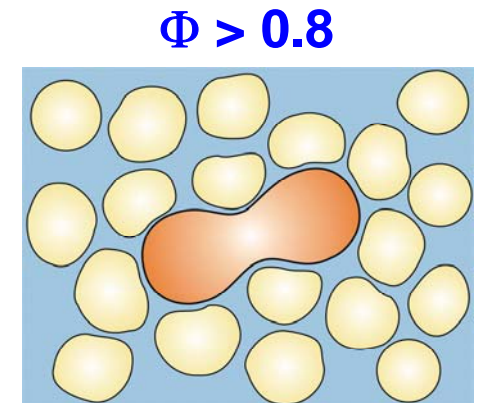
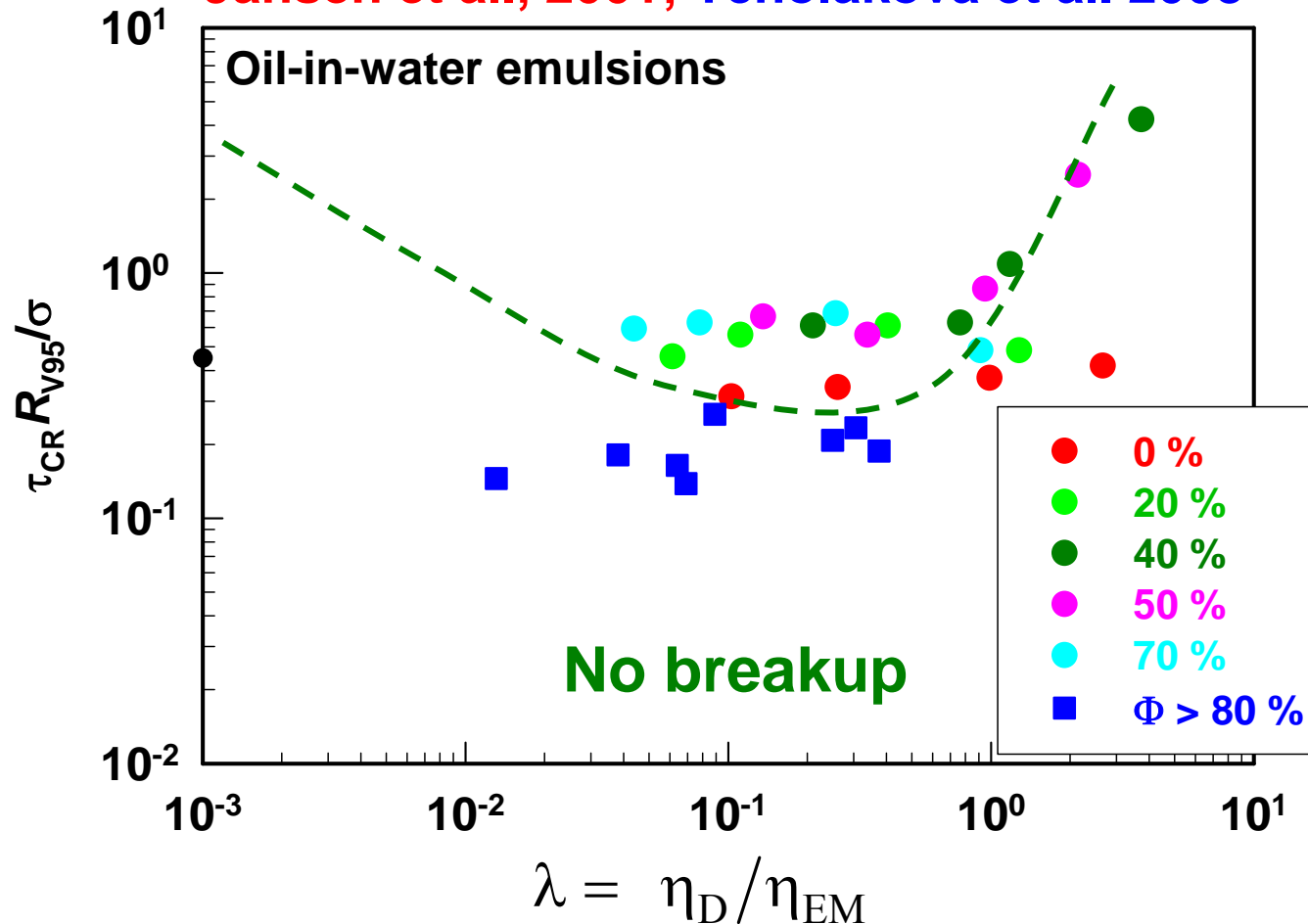
Bentley and Leal, 1986



$$\lambda = \eta_D / \eta_C$$

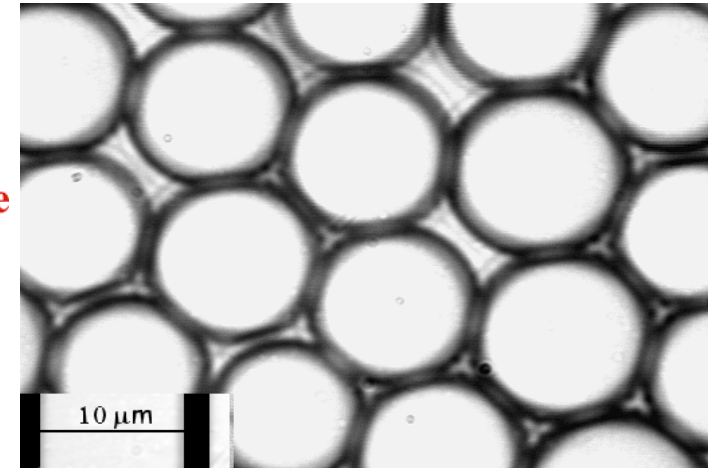
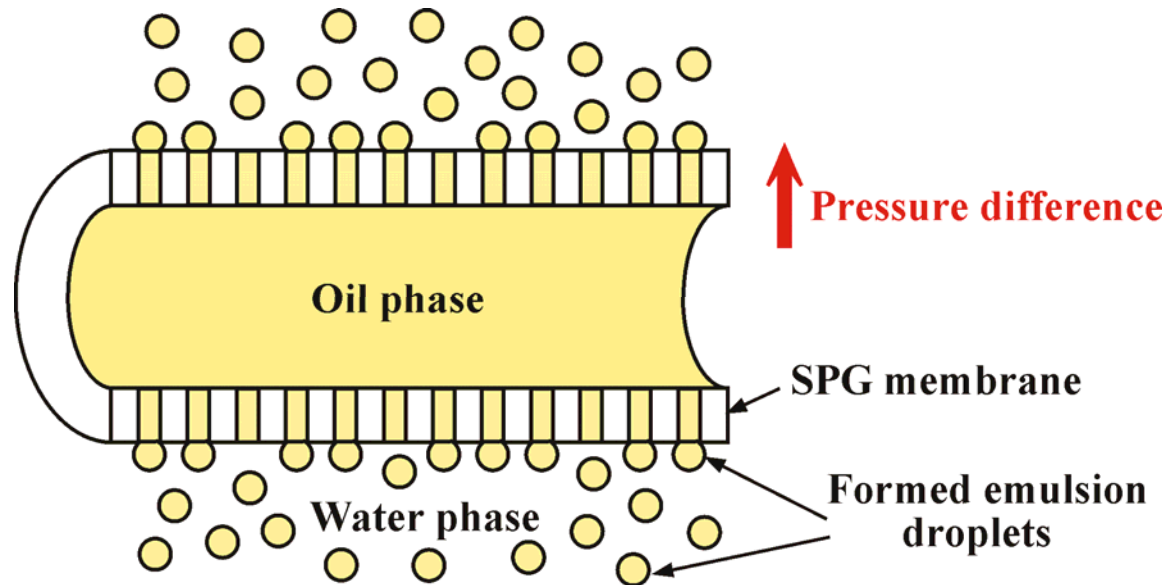
# Drop breakup in concentrated emulsions

Jansen et al., 2001; Tcholakova et al. 2008



Agreement, if emulsion viscosity,  $\eta_{EM}$ , is used instead of  $\eta_c$

# Membrane emulsification



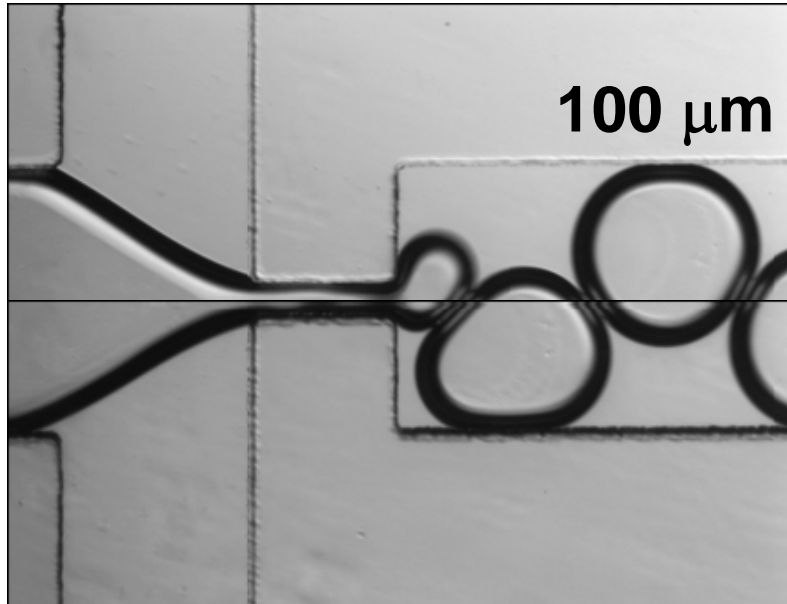
**10 μm monodisperse drops**

**Main factors: Pore size, driving pressure, cross-flow, membrane surface**

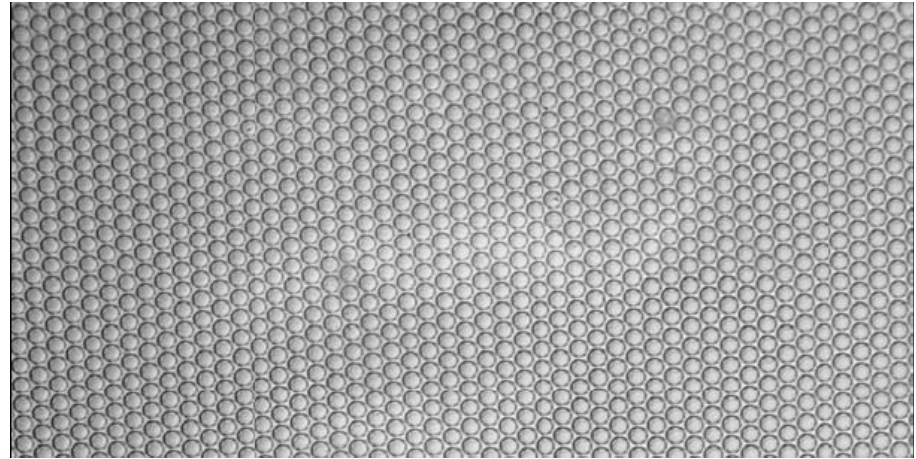
**Main advantage: Monodisperse drops**

**Main disadvantage: Low production capacity**

# Microfluidics



## Monodisperse drops



**Applications: Microreactors, Nanotechnology, Gene engineering...**

**Main factors: Channel size, driving pressure, co-flow**

**Main advantages: Monodisperse drops; Micromanipulation**

**Main disadvantage: Very low production capacity**